Medical Science Sample

Case Report

Acute Appendicitis Masquerading as Distal Intestinal Obstruction Syndrome in Adult

Cystic Fibrosis

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opportunity to assist you with this manuscript. I have edited the text for language, grammar, and improved clarity. As no formatting instructions were provided, I have not looked into this aspect. I have, however, ensured that the style used predominantly by you is consistently maintained throughout the manuscript. Please check your target journal's guidelines and ensu that you comply with all the recommended guidelines Should you have any concerns, please feel free to get back to me.

My best wishes for your success with the manuscript.

Commented [A2]: Please note that the term "masquerading" has not been used elsewhere in the text. Please consider using "mimicking" to be consistent. Overshadowed by Sino-pulmonary infections, With the improved life expectancy in ceystic fFibrosis (CF) patients, there has been an increase in commonly affects gastrointestinal organs manifestations because of secretory and motility dysfunction. Infrequently, these changes can result in dDistal iIntestinal Oobstruction sSyndrome (DIOS), an more and moreincreasingly diagnosed gastrointestinal conditionentity in adult Cystic FibrosisCF patients. We present thea case of a 22-year-old manle who presented to our hospital with right lower quadrant abdominal pain, with Despite the suspicion of acute appendicitis, the patient and was subsequently diagnosed as with DIOS. Our case highlights the importance of considering DIOS as a differential diagnosis of for right lower quadrant abdominal pain in CF patients, especially for by physicians working at community hospitals thatwhich may not have a CEystic Fibrosis care program available.

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1. Introduction

Cystic Ffibrosis (CF) is a genetic disease of that affects multiple organs. With Because of advancementsing in the managementing of CF patients, patients can now often survive become to adulthoods [1]. However, the iImproved life expectancy among adult CF patients has given riseled to an increase in extrapulmonary, notably gastrointestinal, man-ifestations, which did not happen was previously uncommon. Distal Iintestinal oObstruction Ssyndrome (DIOS) continues to be a rising complication in adult CF patients, presenting aswith acute abdominal pain like and mimicking an acute abdominal emergency.

2. Case Report

A 22-year-old Turkish-origin manle with a past medical history of <u>Cystic FibrosisCF</u> presented with a one-day history of right lower quadrant abdominal pain. He described <u>a</u> sharp periumbilical pain that continued to worsen, which then shifted to <u>the</u> right lower quadrant <u>of the</u> abdomen. Prior to the onset of the abdominal pain, he reported <u>experiencing</u> nausea and anorexia for three days. His last bowel movement was two days prior to admission. The patient was also-diagnosed with <u>CFCystic Fibrosis</u> at the age of four, and thehis disease progressed to exocrine

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FIGURE 1: Axial abdominal <u>computed tomographyCAT</u> scan depicting thickening around the terminal ileum and colon (<u>y</u>¥ellow arrows) along with extralumi–nal fluid and reactive lymph nodes.

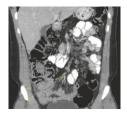


FIGURE 2: Coronal view <u>computed tomography scan</u> with <u>showing</u> thickening of <u>the</u> ileum with <u>a</u> distended appendix (yellow arrows).

measuring 5.3×4.6 mm, and reactive lymph nodes (Figures 1 and 2). Due to extraluminal fluid and cecal wall edema with inflammation, early acute appendicitis could not be excluded as a <u>possible</u> diagnosis. Surgical intervention was <u>performed required</u>, which revealed a ruptured microperforation of a cecal diverticulum and a distended appendix in chronic adhesions, for which he required an appendectomy and partial cecectomy with <u>an</u> intact ileocecal valve (IC valve) valve. Postoperatively, he was diagnosed with DIOS and <u>was</u> subsequently started on pPolyethylene gGly-col. The patient made an unremarkable recovery and was discharged.<u>home to beHe was</u> followed up in the outpatient clinic without and did not have any recurrence of any symptoms.

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3. Discussion

Due to the improved life expectancy of CF patients, DIOS is now being increasingly diagnosed in adult patients with CF. Distal Intestinal Obstruction Syndrome (DIOS) was called a Meconium Ileus-equivalent in the past, described by the collection of viscid fecal material within the lumen combined with sticky mucoid intestinal content adherent to the intesti- nal wall of the terminal ileum and cecum [1]. Perez-Aguilar et al. reported a that the prevalence of DIOS was 19.5% (mean age 20.6 years) among 46 CF patients in a retrospective analysis, while Dray et al. conducted a cross sectional study reporteding a 15.8% (mean age 28.9 years) prevalence in among 171 CF patients in a cross-sectional study [2, 3]. Despite the Though there continues to be a limited assessment of on the prevalence of DIOS in adult CF patients, DIOS is considered more common among adults compared to than among children due to because of increased_disease progression.

Distal Intestinal Obstruction Syndrome (DIOS), previously known as was called a Meconium Ileus-equivalent, in the past, described is characterized by the collection of viscid fecal material within the lumen combined with sticky mucoid intestinal content adherent to the intesti—nal wall of the terminal ileum and cecum [1], Defective intestinal chloride and water secretions into the gut, luminal acidity, and loss of bile salt all contribute to the **Commented [A13]:** Please note, while the discussion section discusses the known facts about the condition, there has been no mention of the present case and how the case relates to the existing literature on the topic. I have therefore included this here to put the case in context to the literature. Please review this addition. Ideally, this discussion of the characteristics of the condition should be included in the Introduction section and this section should discuss this case in the context of previously reported cases.

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Commented [A15]: I believe this sentence would be more appropriate here as this paragraph describes the characteristics and diagnosis of DIOS. development of DIOS [1]. These patients characteristically present with right lower quadrant pain, nausea, abdominal distension, and failure to pass stools or flatus [1, 3]. In some patients, a palpable right lower quadrant mass <u>can may be appreciated present</u> that may be confirmed on abdominal <u>radiographyX-ray</u> [1]. Though abdominal X-rays are radiography is recommended to aid in the diagnosis of DIOS, they are it is inadequate in differentiating ileus from other causes of abdominal pathologies that may present in <u>Cystic FibrosisCF</u> patients [4]. Due to the proximity of the anatomical locationsproximity, as well as the overlapping clinical presentations, appendicitis and intussusception may mimic DIOS. This which further leads to diagnostic uncertainty. Overlap of several intra-abdominal pathologies in CF increases the risk of misdiagnosis, especially with acute appendicitis, as these patient's underlying pathologies may be masked in <u>CF patients</u> with pulmonary infections² using antibiotics, as seen in our case [5, 6].

Osmotic laxatives are the cornerstone of bowel regimens for the treatment of DIOS. The most commonly prescribed <u>laxative</u> is pPolyethylene Gglycol, (PEG) administerrated at a dose of 20–40 ml/<u>kKg/hH</u>, up twith a maximum of 1 <u>lL/kg/h</u> for a total of 8 hours, resulting in aachieving fecal effluent consisting of clear fluid, along with the resolution of abdominal pain and constipation [1, 6]. If the diagnosis remains unclear, and thus, requires surgical intervention, <u>ICileocecal</u> valve resection should be considered to prevent the <u>development and</u> recurrence of intestinal obstruction sequalae-and growth, especially in adolescents [7].

With the increase in immigration of foreigners intothrough America, inner-city and community hospitals may not be sufficiently equipped with a Cystic FibrosisCF care center; moreover, nor may these hospitals may not have programs in provision, with expertise available to other clinicians involved in patient care. Therefore, our case highlights the significance of considering DIOS as a differential diagnosis in CF patients presenting with right lower quadrant abdominal pain, particularly in hospitals without a CF care program available.

Conflicts of Interest

The authors declare that they have no conflicts of interest.

Authors' Contributions

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Life Sciences Sample

Case Report

Methylmalonic Acidemia with Novel MUT Gene Mutations

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A 5-years_old boy presented with poor weight gain and recurrent episodes of fever, feeding problems, and lethargy, sincefrom the age of 11 months, and poor weight gain. He was admitted to our hospital and evaluated for metabolic disorders; subsequently, hecauses and was diagnosed withas methylmalonic acidemia (MMA). He was treated with vitamin B12 and carnitine supplements and has been on-followed -up for the last 3 years. Mutation analysis by next generation sequencing (NGS), supplemented with Sanger sequencing, revealed two novel mutations in exon 5 and 3 of variants in the *MUT* gene responsible for the methylmalonic acidemia (WMA). Recently, he developed dystonic movements, including orofacial dyskinesia. With advent of NGS, judicious use of Thus, next generation sequencing NGS along with Sanger sequencing can help in identification of y causative possibly pathogenic mutations responsible for various clinical conditions and can help in early diagnosis and appropriate treatment of the conditions.

1. Case Presentation

A 5-year-old boy The child presented for the first time at the age of 11 months, presented with recurrent complaints of fever, vomiting, poor feeding, and lethargy since the age of 11 months. On examination We observed that the patient he had pallor and tachypnea and was drowsy. Laboratory tests Further evaluation suggested that the patient had was suggestive of high anion-gap metabolic acidosis with ketonuria (urine ketones_3+) and with-normal electrolytes, blood sugar (94 mg/dLl), vitamin B12, and homocysteine levels. Plasma ammonia and plasma lactate were was-118 units, and plasma lactate was-2.9 units, respectively. Transcranial magnetic stimulation TMS-results wereas normal, but gas chromatography mass spectrometry analysis of but-urine GCMS-revealed elevated 3-OH propionic acid [12.39 retention time (RT)] as well as and elevated methyl malonic acid levels [16.92 RT, Suppl Figure 1, in Supplementary Material available online at https://doi.org/10.1155/2017/8984951]. Since then, their patient child was on a low-protein diet, and carnitine, biotin, thiamine, and vitamin B12 injections:- he Child-was thereafter admitted to the hospital on seven-multiple occasions (7 times) with acute decompensation and managed as per protocol. Mutational analysis was sent for methylmalonic acidemia (MMA) which showed a single heterozygous missense variant c.976 A>G (p.Arg326Gly) in exon 5 of the MUT gene (genomic coordinates: chr 6: 49421405)<u>: as a variant of uncertain significance</u>. Chromosomal microarray analysis done did not reveal any major deletion or duplication that which could disrupt the gene. Since exon 3 and exon 6 were not adequately covered by next generation sequencing (NGS), further evaluation by Sanger sequencing for targeted exons was performeddone, and a second 2nd

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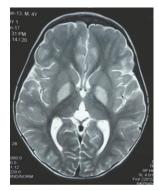
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mutation in exon 3 c.753 G>A (p.=) was identified. The variants were predicted as found to be damaging by the on-SIFT database score (Suppl data) - and as They were also predicted to be deleterious by on-Polyphen-2 and Mutation-Taster, but they were and absent not found in the ExAC database. Brain magnetic resonance image MRI brain of the patient (done at from the age of four4 years) was showeding multifocal cystic encephalomalacic changes with surrounding gliosis in deep white matter predominantly in frontoparietal regions (Figure 1). During In the latest admission of the patient to the hospital, we observed -child was found to have fresh neurological findings in the form of perioral tremors, generalizsed hypertonia, and generalizsed dystonia with clonus with exaggerated deep tendon reflexes. The patient He-was treated with intravenous dextrose and sodium bicarbonate and was continued on carnitine and injection of vitamin B12 injections. Plasma ammonia and plasma lactate were was-18 units and lactate level was 4.9 units, respectively. Brain magnetic resonance image MRI brain of the patientwas repeated and revealed bilateral basal ganglia hyperintensities, suggestive of metabolic stroke. After the subsidence of acute crisis, he was discharged on carnitine, injection of vitamin B12, injections, and trihexyphenidyl. His pParents were counseled regarding the prognosis and for prenatal diagnosis for nextsubsequent pregnancyies.

2. Discussion

MMA presents with lethargy, acidosis, hypoglycemia/ hyperglycemia, ketosis, and recurrent episodes. MMA due to *MUT* gene mutations usually leads to severe phenotypes due to <u>MUT</u> gene mutations, and around 35–40% of cases are due to novelew mutations [1, 2]. There can be <u>Mm</u> issense or nonsense mutations, deletions, insertions, and so on in the <u>MUT</u> gene and so on-can leading to a clinical phenotype.



FIFIGEURE -1: <u>The MRI-Bb</u>rain -magnetic resonance image of <u>in</u> -the -5-year-old boychild -with -MUT-related -<u>methylmalonic acidemiaMMA</u> showing- predominant frontoparietal -abnormalities -in -<u>the</u> form -of encephalomalacia and gliosis.

The advent of NGS technology has enabled better characterization of mutations in several populations. However, Sanger sequencing remains <u>a</u> useful adjunct in molecular testing <u>inof</u> these cases. It is required to find mutations when there is a strong clinical suspicion for them - Sometimes in NGS, due to because of incomplete coverage of the exons by NGS, Sanger sequencing is required to find mutations, if there is strong clinical suspicion. In this study, by

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using both the techniques By careful use of both techniques, we could found ind the two MUT variants responsible for-<u>MMA in the patientthe clinical condition.</u> Previously, iIn a Saudi study on 60 patients of MMA patients, nonsense, missense, and frameshift mutations were detected across the MUT gene [3]. Another study in 43 Chinese patients identified 8 recurrent mutations and 10 novel mutations in the *MUT* gene [4]. A previous Indian study in 15 patients with of clinically diagnosed MMA identified one novel exon 12 mutation in the MUT gene with predicted pathogenicity. In this caseHere, we identified two novel variants, one in exon 3 and another in exon 5 of the MUT gene, and believe by both were labelled as variants of unknown significance-(VUS). The exon 3 variant is a synonymous variant, and a different nucleotide change c.753 G>C (p.Lys251Asn) has been reported earlier in ClinVar. Some synonymous variants can also affect the splicing or protein function and lead to clinical phenotypes. The identified exon 5 variant is novelew, but another close variant c.977 G>A (p.Arg326Lys) has been reported in ClinVar. The variants were found to be deleterious on bioinformatic analysis and were absent not found in the ExAC database. Both variants identified in the present case could be responsible for possibly explain the phenotype of MMA phenotype in the child. MUT-related MMA has poor prognosis in most cases. Specializsed diet and supplements may not improve the outcomess, even if MMA is diagnosed early. Early recognition and appropriate treatment of acute crises are necessary. Metabolic stroke can sometimes occur in the absence of acute metabolic decompensation: thus, , so-meticulous neurological examination at everyeach visit is importantuseful. The treatment options for MMA for therapy include early liver transplantation [5]; and possibly gene therapy could also be used in the future. Genetic counseling and prenatal diagnosis could help the se-families of the patients in making reproductive decisions in the future.

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Conflicts of Interest

The authors declare that they have no conflicts of interest.

Acknowledgments

The authors would like to acknowledge Dhiti Omics Tech–nologies Pvt Ltd for <u>assistance</u> help-in mutation analysis.

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Physical Science Sample

 Structural
 Prediction
 of
 Bisbis(di-p-anisole)-1,4-azabutadienebis[triphenylphosphine]ruthenium(II)

 bis[triphenylphosphine]ruthenium(II)
 complex
 Using using ³¹P NMR Spectroscopyspectroscopy

Author Details

Abstract¹

In this study. The present paper reports the use of ³⁴P NMR spectroscopy to predict the isomer structures of [bis_4-methoxy-phenyl-[3-(4-methoxy-phenyl)-allylidene]-amino]-bis[triphenylphosphine]ruthenium(II) complex, also known as bis(di-p-anisole)-1,4-azabutadiene-bis[triphenylphosphine]ruthenium(II)₅ complex, was synthesized usinges. The complexation reaction was earried out (di-p-anisole)-1,4-azabutadiene (compound 1), triphenylphosphine, and ruthenium chloride in 2:2:1 ratio under refluxing conditions] of (di-p-anisole)-1,4-azabutadiene (compound 1), triphenylphosphine (PPh₂), and ruthenium chloride in the ratio of 2:2:1 ratio under refluxing conditions] of (di-p-anisole)-1,4-azabutadiene (compound 1), triphenylphosphine (PPh₂), and ruthenium chloride in the ratio of 2:2:1 ratio five 5 hhours. The formation of the In addition, ruthenium(II) complex were was confirmed by also characterized using FTIR and UV_-Vis spectroscopic__analysesto_support the formation of ruthenium(II) complexes. ³¹P NMR spectroscopy pie study on ruthenium(II) complexes suggested indicated the presence of that there are three isomers present after the complexation reaction.

Keywords: ^{[51}P NMR spectroscopy; FTIR spectroscopy; UV–Vis spectroscopy; Ru complex; Isomers; Structure prediction

¹NMR, nuclear magnetic resonance; FTIR, Fourier transform infrared; UV–Vis, ultraviolet–visible

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https://www.elsevier.com/journals/inorganic-chemistrycommunications/1387-7003/guide-for-authors#25000 In the cases where additional information is required from you, I have added comments to bring them to your attention.

Should you have any questions, please feel free to get back to me.

My best wishes for your success with the manuscript.

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1. Introduction

Nuclear magnetic resonance (NMR) spectroscopy is an essential instrument analytical tool in the field of chemistry as it can helps determine elucidate the structure of a molecule, identify detect the presence of impurities in a sample, and determine the rates of formation and as well as degradation of a compound-Even in 1970s, NMR has was used as early as in the 1970s already been used to determine detect the cancer formation which and was identified to be offered a simple, fast, and low-cost method to for this purpose identify cancer formation [1-3].

For Inorganic inorganic chemists commonly use, using of ³¹P NMR spectroscopy to identify the structures of a complexes containing phosphine ligands is very common [4, 5]. ThFor example, the e well-known examples is the use of ³¹P NMR spectroscopy to determine the mechanism of Wilkinson hydrogenation was determined by ³¹P NMR spectroscopy, mechanism based on by identifying the coupling patterns among the phosphine ligands as well as those and also the coupling constants between the phosphine ligands as well as and the rhodium(I) metal centrecenter [6].

In <u>As part of</u> our long-term research interest on the synthesis of in-ruthenium(II) complexes-synthesis, we used the (di-*p*-anisole)-1,4-azabutadiene (1) and_triphenylphosphine (PPh₃) as the ligands to for reaction react with ruthenium trichloride under reflux conditions. The structures of the Products-resulting complexes were formed, were checked-identified by using ³¹P NMR spectroscopy, FTIR spectroscopy, and UV-Vis spectroscopy and the results found in the spectra are worth to be discussed in the present communication. For inorganic chemist, using of ³⁴P NMR to identify the structure of a complex containing phosphine ligands is very common [4, 5]. The well-known examples is the use of ³⁴P NMR spectroscopy to determine the Wilkinson hydrogenation mechanism by identifying the coupling patterns among phosphine ligands and also the coupling constants between phosphine ligands as well as rhodium(I) metal centre [6].

2. Methodology

The ruthenium complexes were characterized using UV/Vis, FTIR, and ³¹P NMR spectroscopy. The IR spectra were recorded <u>using on</u> a Thermo Scientific Nicolet iS10 <u>spectrophotometer in using KBr</u> discs. The ¹H NMR spectrum for of compound 1 and ³¹P NMR spectrum for of the ruthenium(II) complexes were recorded <u>using on a</u> JEOL JNM-ECA 500 spectrometer with TMS as <u>an-the</u> internal standard. The absorption spectra waswere recorded with on a Jasco V-630 <u>UV-Vis</u> spectrophotometer.

2.1. Preparation of (4-Methoxy-phenyl)-[3-(4-methoxy-phenyl)-allylidene]-amine or (dip-Anisole)-1,4azabutadiene (1)

4-Methoxycinnamaldehyde (1.62 g, 10.00 mmol) was dissolved in 10 mL of ethanol, <u>and</u>-followed by the <u>addition of</u> 4-methoxyaniline (1.23 g, 10.00 mmol) which was then added to solution. The rReaction mixture was stirred <u>and to obtain a resulted in green-yellow solid</u>, <u>which</u>. The solid-was filtered, washed with 5 mL of ethanol, and dried *in vacuo*. The solid was purified by dissolving it in DCM and then layered with hexane via slow diffusion to yield compound 1.: Yyield: 2.368 g (88.7%); IR (KBr, cm⁻¹)_v: 3036 (C-H stretching), 1627 (C=N- stretching), 1601 (C=C stretching, aliphatic), 1575 and 1468 (C=C stretching, aromatic), and 1110 (OCH₃ stretching); ¹H NMR (500 MHz, CDCl₃) *i*: 8.25 (d, 1H, Hz, -CH=N-), 7.47 (d, 2H, Hz₇), 7.18 (d, 2H, Hz₇), 7.05 (t, 1H, Hz, H-C_a), 6.99 (m, 1H, H-C_β), 6.90 (d, 4H, Hz₇), 3.83 (s, 3H, OCH₃), and 3.81 (s, 3H, OCH₃); UV_-Vis (DCM, /nm): 273, 373; Anal. Calc. for C₁₇H₁₇O₂N (%): C, 76.38; H, 6.41; N, 5.24; Found (%): C, 76.75; H, 6.31; N, 5.05.

2.2. Preparation of [Bis4-methoxy-phenyl-[3 (4-methoxy-phenyl)-allylidene]-amino}]-bis-[triphenylphosphate]ruthenium(II) or Bis(di-p-anisole)-1,4-azabutadiene}bis[triphenylphosphine]ruthenium(II) Complexes

For the synthesis of bis(di-p-anisole)-1,4-azabutadiene-bis[triphenylphosphine]ruthenium(II) complex, RuCl₃·xH₂O (2.070 g, 1.0 mmol) and PPh₃ (0.525 g, 2.0 mmol) were added to a round_bottom flask containing 10 mL ethanol_ and tThe mixture was then refluxed. Compound 1 (0.316 g, 2.0 mmol) was then added to the round bottom flask, and the mixture was refluxed again. The resulting pPale_maroon solids were wasformed_, filtered and washed with hexane, and the p. Precipitate was dried *in vacuo*: IR (KBr, cm⁻¹) v: 3034 (C-H stretching), 1661 (C=N), 1576 (-merged IR band of for aliphatic and aromatic C=C stretching from aliphatic and aromatic), 1469 (C=C stretching of aromatic ring), and 654 (Ru-C), and 577 **Commented** [A10]: I have deleted the section heading as the Introduction, Experimental, Results and Discussion sections should be combined into a single untitled section.

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$(\text{Ru-N})_{::}^{:31}$ P NMR (202.5 MHz, CDCl₃) δ : 49.7 (d, 1P, Hz), 47.4 (d, 1P, Hz), 41.7 (d, 1P, Hz), 39.7 (d, 1P, Hz), 35.1 (s, Ph₃P=O), and 29.9 (s, 1P); UV_-Vis (DCM) (): 321 and 382.

On the other hand, the <u>The</u> binding of compound <u>1</u> to <u>the</u> ruthenium(II) metal <u>centrecenter</u> <u>wasis_can_be</u> confirmed using FTIR and UV_-Vis <u>spectroscopyspectroscopies</u>. The <u>Comparing the IR</u> <u>spectra spectrum</u> <u>between of compound <u>1</u> and <u>the</u> ruthenium complexes (Figure Fig. <u>41</u>) reveals that the <u>,</u> the vibrations of C=N and C=C stretching <u>bands are shifted</u> with respect to those in <u>1</u>, thereby <u>bands have been shifted</u> confirming the <u>after</u> binding <u>of <u>1</u> to the ruthenium(II) metal <u>centrecenter</u>. The <u>For</u> C=N stretching <u>bands is</u> <u>blue_, it shifted</u> from 1627 cm⁻¹ in the spectrum of <u>compound <u>1</u> to 1661 cm⁻¹ in the spectrum of the ruthenium complex [97, 108]. In <u>contrast</u>, <u>whereas the for C=C</u> stretching, the IR band appears at 1601 cm⁻¹ in the spectrum of <u>compound <u>1</u> but it is notcannot be clearly shown detected in the spectrum of the complex, because the <u>IR bands of aliphatic and aromatic</u> C=C bands for <u>aliphatic and aromatic</u> were merg<u>eing</u> into one <u>a single</u> broard IR-band <u>centredcentered</u> at 1576 cm⁻¹. Nevertheless, <u>the two</u> additional IR-peaks are present <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the <u>spectrum</u> <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the spectrum <u>at 577</u> and 654 cm⁻¹ in the finger-print region of the</u></u></u></u>

FigureFig. 41: IR spectra of (a) compound 1 (a) and (b) ruthenium(II-) complexes (b).

The complexation of compound 1 to the ruthenium(II-) metal centrecenter is can be further supported by the UV_- \underline{v} is data spectra as shown in Figure Fig. 52. For In the case of compound 1, two absorption bands were are observed at 273 and 372 nm, which are assigned to the transitions of in the benzene ring and transition of thei-imine group [4210], respectively. After the complexation, both the absorption bands undergo significant bathochromic shifts to shifts to 321 and 382 nm, respectively, thereby -confirming the Significant shifts of these two absorption bands have proven compound 1 was successfully bound binding of 1 to the ruthenium(II) metal centrecenter via the nitrogen atom from in the C=N group and the carbon atom from in the aliphatic C=C aliphatic group in of the C=C-C=N moiety.

FigureFig. 52: UV_-Vis spectra of (a) compound 1 (b) ruthenium (II) complex (b).

3. Results and Discussion

Characterization of the ruthenium complexes was done using UV/Vis, FTIR, and ³⁴P NMR spectroscopy. The IR spectra was found by Thermo Scientific Nicolet iS10 in KBr dise. ⁴H NMR spectrum for compound 1 and ³⁴P NMR spectrum for ruthenium(II) complexes obtained through JEOL JNM-ECA 500 spectrometer with TMS as an internal standard. The absorption spectra recorded with Jasco V-630 spectrophotometer.

Once the complexation was confirmed, as discussed above, The-the ³¹P NMR spectrum of the ruthenium complex (Fig. 3) was analyzed for its detailed structural elucidation. The ³¹P NMR spectrum of the product shows appearance of two pairs of doublets and one singlet, indicating in the ³⁴P NMR spectrum for ruthenium complexes (Figure 1) indicate the formation of that there are three isomers (1:1:1 ratio) present in as a result of the-the complexation reaction with the ratio of 1:1:1.

FigureFig. 13: ³¹P NMR spectrum for of ruthenium(II) complexes.

The singlet at 29.88 ppm reveals that the two PPh₃ <u>units</u> are magnetically equivalent in <u>the</u> nuthenium(II) complex. There can be three possible structures based on this singlet. In the first case, the <u>In this case</u>, the two PPh₃ <u>units</u> are <u>either</u>-located at <u>the</u> axial positions <u>of an octahedron</u> and <u>are</u>, which is trans to each other (FigureFig. 24(a)) [117], while in the other two cases, they are located on the <u>or located at equatorial plane</u>, which is only trans only to <u>either one of the</u> C atoms from in the C=C bond (Fig. 4(b)) or the N atom from in the N=C____bond (FigureFig. 24(bc)).

FigureFig. 24: Postulated structures of (a) *trans-* and ((b) and (c)) *cis-*[bis(di-*p*-anisole)-1,4-azabutadiene}]-bis[triphenylphosphine]ruthenium(II).

<u>Meanwhile, a The</u> pair of doublets at 41.84 and 39.74 ppm with <u>a</u> -coupling constant of 21 Hz is assigned to <u>a the</u> cis_isomer of <u>the</u> ruthenium(II) complex <u>as</u> shown in <u>FigureFig. 35(a)</u>. <u>Lastly The other</u>, <u>another</u> pair of doublets at 49.80 and 47.36 ppm with <u>a</u> coupling constant of 38 Hz is assigned to <u>a the</u> trans_ruthenium(II)

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complex shown in (FigureFig. 35(b)). It is evident that the The difference in the coupling constants between of the ruthenium(II) complexes arisesin Figures 3(a) and 3(b) is due owing to the positions of the PPh₃ ligands. The doublet with a smaller coupling constant (t, namely, 21 Hz) is, was assigned to the cis_isomer because both the PPh₃ ligands are in the equatorial plane. The The presence of doublets originate because the _for the PPh₃ ligands in the complex is shown in Figure 3(a) because both PPh₃ ligands are trans to different atoms, that is, (nitrogen and carbon) atoms. For In the ruthenium(II) complex shown in shown in FigureFig. 35(b), the two PPh₃ ligands are located at the axial positions and are trans to each other. The Lastly, the single peak observed at 35.14 ppm is attributed to the presence of the triphenylphosphine oxide [128].

FigureFig. 35: Postulated structures of (a) *cis*- and (b) *trans*-[bis(di-*p*-anisole)-1,4-azabutadiene]}-bis[triphenylphosphine]ruthenium(II)].

On the other hand, the binding of compound 1 to ruthenium(II) metal centre can be confirmed using FTIR and UV Vis spectroscopy. Comparing the IR spectra between compound 1 and ruthenium complexes (Figure 4), the vibrations of C-N and C-C stretching bands have been shifted after binding to ruthenium(II) metal centre. For C-N stretching band, it shifted from 1627 cm⁻¹ in compound 1 to 1661 cm⁻¹ in ruthenium complex [9, 10], whereas for C-C stretching, the IR band appears at 1601 cm⁻¹ in **compound 1** but it is not clearly shown in the complex because the IR bands of C-C bands for aliphatic and aromatic were merging into one board IR band centred at 1576 cm⁻¹. Nevertheless two additional IR peaks are present in the finger print region at 577 and 654 cm⁻¹ indicating the formation of respective Ru N and Ru C bonds [11]. Figure 4: IR spectra of compound 1 (a) and ruthenium(II) complexes (b).

The complexation of compound 1 to ruthenium(II) metal centre can be further supported by the UV-vis data as shown in Figure 5. For compound 1, two absorption bands were observed at 273 and 372 nm which are assigned to transition of the benzene ring and transition of the imine group [12], respectively. After the complexation, both absorption bands shifts to 321 and 382 nm, respectively. Significant shifts of these two absorption bands have proven compound 1 was successfully bound to ruthenium(II) metal centre via the nitrogen atom from C=N group and carbon atom from C=C aliphatic group in C=C C=N moiety. The bathochromic shift of these two absorption bands was due to the backbonding of electrons from Ru to the antibonding orbitals of C=C-C=N moiety in compound 1. This, in turn, has weakened the bond in C=C C=N [13].

Figure 5: UV-Vis spectra of compound 1 (a) and ruthenium (II) complex (b).

In addition, the data from IR and UV-Vis revealed tThe successful binding of at compound 1 has bound to the ruthenium(II) metal centrecenter was confirmed from the IR and UV-Vis spectral data.

The The ³¹P NMR spectra revealed the evidence from ³⁴P NMR spectrum has shown the presence of three isomers of the bis(di-*p*-anisole)-1,4-azabutadiene}-bis[triphenylphosphine]ruthenium(II) complex in the 1:1:1 ratio-of 1:1:1.1 [Two of the three isomers are those shown in Fig. 5, i.e., one cis and one trans isomer, while the third isomer could be any one of those shown in Fig. 4. In addition, the data from IR and UV-Vis revealed that compound 1 has bound to ruthenium(II) metal centre.

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15.

HIGHLIGHTS

- Three isomers were detected for a phosphine-bearing Ru complex using ³¹P NMR.
- Formation of Ru-N and Ru-C bonds were confirmed by FTIR spectroscopy.
- At least one cis isomer and one trans isomer of the complex were formed.

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